liquid solids							noble gases				Name: lanthanides				
	alkali metals					metalloids					actir	nides			
	alkaline earth metals					non-metals					13	14	15	16	17
2 3 4 Be	halogens					metals (leave white)					5 B	6 C	7	8	9 10
3 11 12 Na Mg	3	4	5	6	7	8	9	10 10 ans	11	12	13 Al	14 Si	15 P	16 S	17 18
4 19 20 K Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 36
5 37 38 Rb Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 54 I
6 55 56 Cs Ba	*57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80	81 Tl	82 Pb	83 Bi	84 Po	85 At 86
7 87 88 Fr Ra	**89 Ac	104 Rf	105 Db	106 Sg	107 Ns	108 Hs	109 Mt	110 Ds	111 Rg						
	*58 Ce **90 Th	59 Pr 91 Pa	60 Nd 92 U	61 Pm 93 Np	62 Sm 94 Pu	63 Eu 95 Am	64 Gd 96 Cm	65 Tb 97 Bk	66 Dy 98 Cf	67 Ho 99 Es	68 Er 100 Fm	69 Tm 101 Md	70 Yb 102 No	71 Lu 103 Lr	

Coloring The Periodic Table

You will need to use at least 10 different markers/pencils: 2 thin dark markers, 4 light colored pencils or highlighters and 4 thick markers any color.

1. The two elements that are liquids at room temperature are bromine (35) and mercury (80). *Write their symbols* (**Br**, **Hg**) using the same color of *thin dark marker* in the appropriate spaces. Also, color in the letters of the word *liquid* with this color



_2. The elements that are gases at room temperature are **H**, **N**, **O**, **F**, **Cl** and the six elements in the last group (column) **He**, **Ne**, **Ar**, **Kr**, **Xe**, **Rn**. Use a different color of *thin dark* marker than the color you used for the liquids, and w*rite their symbols in the appropriate spaces*. Also, color in the letters of the word *gas* with this color.

All of the rest of the elements are solids at room temperature. Their colors are already black

*3. Color the *border around the symbols for the following. Use a different colored thick marker for each group. Also color the appropriate border in the "key" at the top of the table.

*Color border only *Color border only —A The alkali metals include Li, Na, K, Rb, Cs and Fr in Group 1 —B. The alkaline earth metals include Be, Mg, Ca, Sr, Ba and Ra in Group 2 —C. The halogens include F, Cl, Br, I and At Group 17 —D. The noble gases include He, Ne, Ar, Kr, Xe and Rn in Group 18

_**4. The metalloids are **B**, **Si**, **As**, **Te**, **At**, **Ge**, **Se** and **Sb**. Color the ***inside of their blocks* **Color the a *light colored pencil* you haven't used yet *Also color the appropriate box in the* inside only *"key" at the top of the table.*



____5. The nonmetals are *all* of the elements to the *right* of the metalloids (including the halogens and noble gases). *H* is also a nonmetal Color the ***inside of the blocks* using a *light colored pencil* you haven't used yet. There should be 16 blocks colored this way. *Also color the appropriate box in the "key" at the top of the table.*

All of the rest of the elements to the <u>left</u> of the metalloids are <u>metals</u>. You do not need to color these blocks.

6. Color a *diagonal line through the symbols* for the following with a *light color* marker or pencil. *Use a different color for each group. Also color the appropriate box in the "key" at the top of the table.*

