Name:	Pd.	Date:	

Periodic Table Facts Worksheet

Use the following website for help: http://www.chem4kids.com/files/elem_pertable.html

1.	Horizontal rows are called
2.	Elements in the same (row) are filling up the same energy level as you move from left to right.
3.	Vertical columns are called
4.	electrons are the electrons in the outermost shell of the atom.
5.	All families have the same number of electrons.
6.	The have full outer electron shells which makes them inert or nonreactive.
7.	Noble gases are the elements in column (includes helium, neon,, and)
8.	Noble gases are elements that have a valence electron shell.
9.	Noble gases are also called gases because they don't react with other elements.
10	.The family are the most active metals.
11	. Alkali metals are the elements in the column.
12	is NOT an alkali metal because it behaves unlike any other element and does not belong to a family. It is our only orphan!
13	.The family are the most active NONmetals.
14	. Halogens are elements in the column (includes fluorine, chlorine,
15	. What do you notice about the number of valence electrons as you move from left to right across a row or period in the periodic table? (Na ® Mg® Al® Si® P ®S ®Cl® Ar)