Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Pd. \_\_\_\_\_ Date: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**Periodic Table Review**

***Directions***: Identify each statement as True or False based on the underlined information. If False, correct the underlined part.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ 1. Alkaline Earth Metals are **group 1** and are found to the left of the zig-zag line.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_2. **Noble gases** are in group 18 and are found to the right of the zig-zag line.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_3. Halogens and Noble gases are both **non-metals**.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_4. **Metalloids** are ductile and malleable.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_5. Alkali Metals have a luster and are found in **group 2**.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_6. **Rows** on the periodic table are called groups or families.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_7. Being malleable, ductile, and having a luster are all **properties of non-metals**.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_8. The **transition metals** include groups 3-12 and are found between the Alkaline Earth metals and the metalloids.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_9. Halogens, group 17, are a family of the most **non-reactive** non-metals on the periodic table.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_10. There are seven **groups or families** on the periodic table.

Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Pd. \_\_\_\_\_ Date: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**Periodic Table Review**

***Directions***: Identify each statement as True or False based on the underlined information. If False, correct the underlined part.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ 1. Alkaline Earth Metals are **group 1** and are found to the left of the zig-zag line.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_2. **Noble gases** are in group 18 and are found to the right of the zig-zag line.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_3. Halogens and Noble gases are both **non-metals**.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_4. **Metalloids** are ductile and malleable.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_5. Alkali Metals have a luster and are found in **group 2**.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_6. **Rows** on the periodic table are called groups or families.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_7. Being malleable, ductile, and having a luster are all **properties of non-metals**.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_8. The **transition metals** include groups 3-12 and are found between the Alkaline Earth metals and the metalloids.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_9. Halogens, group 17, are a family of the most **non-reactive** non-metals on the periodic table.

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_10. There are seven **groups or families** on the periodic table.