

Name: \_\_\_\_\_ Pd. \_\_\_\_\_ Date: \_\_\_\_\_

### **Periodic Table Review**

**Directions:** Identify each statement as True or False based on the underlined information. If False, correct the underlined part.

- \_\_\_\_\_ 1. Alkaline Earth Metals are **group 1** and are found to the left of the zig-zag line.
- \_\_\_\_\_ 2. **Noble gases** are in group 18 and are found to the right of the zig-zag line.
- \_\_\_\_\_ 3. Halogens and Noble gases are both **non-metals**.
- \_\_\_\_\_ 4. **Metalloids** are ductile and malleable.
- \_\_\_\_\_ 5. Alkali Metals have a luster and are found in **group 2**.
- \_\_\_\_\_ 6. **Rows** on the periodic table are called groups or families.
- \_\_\_\_\_ 7. Being malleable, ductile, and having a luster are all **properties of non-metals**.
- \_\_\_\_\_ 8. The **transition metals** include groups 3-12 and are found between the Alkaline Earth metals and the metalloids.
- \_\_\_\_\_ 9. Halogens, group 17, are a family of the most **non-reactive** non-metals on the periodic table.
- \_\_\_\_\_ 10. There are seven **groups or families** on the periodic table.

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